

# Micro-bioreactors Controlled with Photonic Ionogel Actuators

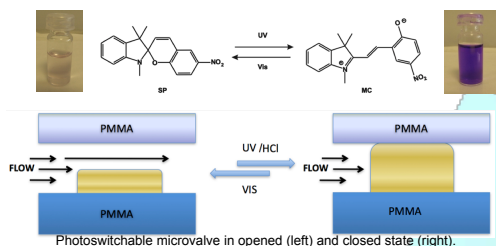
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## 1. Abstract

In the recent years, advances in micro-fluidic techniques for environmental applications have brought wide opportunities for improving of the capacity to monitor water quality. However, the development of fully integrated micro-fluidic devices capable of performing complex functions requires the integration of micro-valve with appropriate performance, since they are essential tools for the control and manipulation of flows in micro-channels.[1] Ionogels with incorporated spiropyran can be used as valves by photopolymerizing the gels in certain shapes. Depending on the ionic liquid, ionogels give the possibility of tuning several micro-valve actuation times and so independently control liquid flows within the channels under a common illumination source.[2]

## 2. Introduction

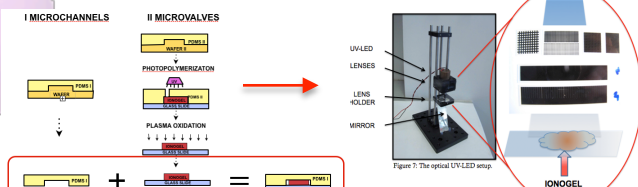
- Ionogel containing spiropyran expands when immersed in HCl and contracts with white light back, and due to this property it can be used as a valve to block microchannel as shown below.
- Non invasive stimuli such as light offers improvements in versatility using manifold fabrication and control of actuation mechanism.



## 3. Methods

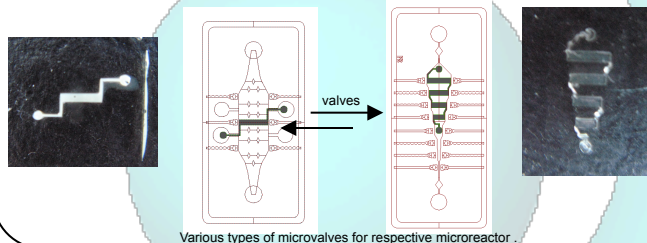
### Fabrication of microstructures and microreactors

Fabrication of microstructures with three different ionic liquids:  
 1.  $[P_{6,6,6,14}][dca]$     2.  $[P_{6,6,6,14}][Cl]$     3.  $[P_{6,6,6,14}][NTf_2]$



Fabrication of microreactors (left) and set up for photopolymerisation of microstructures. (right)

### Microreactors with microvalves

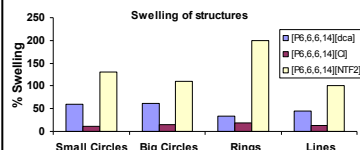


## 4. Results

### Swelling and shrinking of microstructures



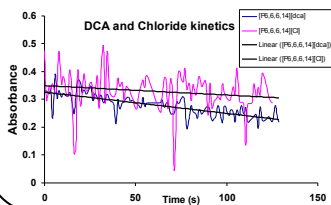
Ionogel in the initial state (left), after 2 hours in HCl (centre) after white light irradiation (right).



Ionic Liquid	% Shrinking after 15 mins
$[P_{6,6,6,14}][DCA]$	31.8
$[P_{6,6,6,14}][Cl]$	6.1
$[P_{6,6,6,14}][NTF_2]$	95.8

Swelling of microstructures in different ionogels (left) and shrinking of these after 15 min (right)

### Kinetics of opening the microvalve



$$[P_{666,14}][dca] K = 0.0008$$

$$[P_{666,14}][Cl] K = 0.0003$$

1<sup>st</sup> order kinetics for  $[P_{666,14}][dca]$  and  $[P_{666,14}][Cl]$  (left) and kinetic values (above)

## 5. Conclusion

- Successful fabrication and actuation of photoswitchable microvalves.
- Greatest height change obtained for  $[P_{6,6,6,14}][NTF_2]$  with 95.83 %.
- Fastest shrinking obtained for  $[P_{6,6,6,14}][NTF_2]$  with 95.8% in 15 mins.
- Successful fabrication of microbioreactors and microvalves.
- Initial kinetics results show the faster response of the microvalve with  $[P_{6,6,6,14}][dca]$ , with kinetic constant of  $k=0.0008$ .

## 6. Future Work

1. Carry out the kinetic tests for ionogel with  $[P_{6,6,6,14}][NTF_2]$  and control tests using UV-Vis spectroscopy.
2. Test the effect of different concentrations of HCl on the kinetics of the valves.
3. Test the performance of the microreactor with dye solutions.

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[1] M. Czugala et al., Proceedings SPIE 8107, 81070C, (2011); doi:10.1117/12.895330. [2] F. Benito-Lopez et al. Lab Chip 10, (2010), 195-201